

$J$  is thus the determinant of the Jacobian of the transformation, or the "functional determinant."

The strain,  $N_{jk}$ , is defined, somewhat arbitrarily, from the difference in the squares of the lengths of line elements by:

$$2N_{jk} da_j da_k = dx_i dx_i - da_i da_i \quad (2.15)$$

$$N_{jk} = 1/2 \left( \frac{\partial x_i}{\partial a_j} \frac{\partial x_i}{\partial a_k} - \delta_{jk} \right)$$

Here and in the following the Einstein summation convention for repeated subscripts applies.  $\delta_{jk}$  is the Kronecker delta.

Expanding the internal (strain) energy in a power series in the strains, one obtains (at constant entropy):

$$\begin{aligned} \rho_0 [E(N,S) - E(0,S)] = & 1/2 c_{ijkl}^S N_{ij} N_{kl} + 1/6 c_{ijklmn} N_{ij} N_{kl} N_{mn} \\ & + 1/24 c_{ijklmnpq}^S N_{ij} N_{kl} N_{mn} N_{pq} + \dots \end{aligned} \quad (2.16)$$

In this expression the  $c_{ijkl}^S \dots$ , represent the second and higher order isentropic elastic stiffness coefficients. The first-order term is missing since the reference state is considered to be one of zero stress and strain.

We now define quantities, called thermodynamic tensions, by

$$t_{ij} = \rho_0 \left( \frac{\partial E}{\partial N_{ij}} \right)_S \quad (2.17)$$

In terms of these quantities the elastic constants are

$$c_{ijkl}^S = \left( \frac{\partial t_{ij}}{\partial N_{kl}} \right)_S = \frac{\partial^2 E}{\partial N_{ij} \partial N_{kl}}$$

and similarly for the higher order coefficients. Consequently,

$$\rho_0 dE = t_{ij} dN_{ij} \quad (dS = 0)$$

Finally, the equilibrium (non-dissipative) components of the stress are obtained from the thermodynamic tensions by the relations,

$$\sigma_{km} = (1/j) \frac{\partial x_k}{\partial a_j} \frac{\partial x_m}{\partial a_i} t_{ij} \quad (2.18)$$

The above formulas provide isentropic constitutive relations in terms of the elastic stiffness coefficients. Other forms of constitutive relations can, of course, be derived in a similar fashion.

Low pressure acoustic measurements yield a mixed third-order constant of the form:

$$C_{ijkmpq} = \left( \frac{\partial c_{ijkm}^S}{\partial N_{pq}} \right)_T$$

where the subscript T means the derivative is taken at constant temperature.

The corresponding purely isentropic constant is given by:

$$C_{ijkmpq}^S = C_{ijkmpq} + (T/\rho_0 C_t) c_{kmpq}^S \alpha_{uv} [C_{ijkmrs} \alpha_{rs} - \left( \frac{\partial c_{ijkm}^S}{\partial T} \right)_t] \quad (2.19)$$

where  $C_t$  is the specific heat at constant tension and the  $\alpha_{uv}$  are thermal expansion coefficients,

$$\alpha_{uv} = \left( \frac{\partial N_{uv}}{\partial T} \right)_t$$

In view of the symmetry of the stress and strain tensors, the number of subscripts can be reduced by adopting the following convention:

$$\begin{array}{ll} 11 \rightarrow 1 & 32 \rightarrow 4 \\ 22 \rightarrow 2 & 31 \rightarrow 5 \\ 33 \rightarrow 3 & 21 \rightarrow 6 \end{array}$$

This convention is employed in the following.